²⁷Al MAS NMR SPECTROSCOPY STUDY OF Eu²⁺-DOPED AND Dy³⁺-CO-DOPED SrAl₄O₇

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The Eu²+-doped strontium aluminate $SrAl_4O_7$ samples have shown the blue-green persistent luminescence at 490 nm while the co-doping with Dy^{3+} shifts the maximum of emission to 475 nm. Undoped, 3% Eu-doped and 6% Dy-co-doped $SrAl_4O_7$ samples were prepared by the solid state-reaction method and studied by the solid-state ^{27}Al MAS NMR applying the single pulse-acquire and Hahn-echo pulse sequences. It was shown that the Eu^{2+} with Dy^{3+} ion doping did not affect the bulk structure as well as the local Al environment in $SrAl_4O_7$. This means that large shifts of the emission maximum cannot be caused by changes in the local environment upon the co-doping of $SrAl_4O_7$: Eu^{2+} with Dy^{3+} . However, the spectral features observed in the range between the signals of 4- and 6-coordinated Al (20–40 ppm) indicate that certain phase imperfections are present in all studied samples, and most probably amorphous/glassy domains were formed. Note that such amount of phase impurities was not detected by standard XRD or FTIR methods. This has revealed the ^{27}Al MAS NMR technique to be a very effective tool monitoring the phase perfectness in series of strontium aluminate samples.

Keywords: solid-state NMR, magic-angle spinning, strontium aluminate, europium, dysprosium **PACS:** 61.72.Hh, 76.60-k, 82.56.-b

1. Introduction

Strontium aluminates, doped with rare-earth metal ions, have been studied for a long time for their excellent properties, such as a high quantum efficiency and a long persistence of phosphorescence, as well as due to potential industrial applications in fluorescent lamps, plasma displays, light emitting diodes, etc. [1–3]. Beside a very high quantum efficiency and a long persistence luminescence, strontium aluminates doped with Eu²⁺ have a better stability than other alkaline earth aluminates [4]. Several different strontium aluminate phases are well known, e.g. $SrAl_2O_4$, $Sr_4Al_{14}O_{25}$, $SrAl_{12}O_{19}$ and $Sr_3Al_2O_6$ [5]. However, a particular one – $SrAl_4O_7$ – was much less in-

vestigated, probably due to the difficulties arising at the preparation of monophasic samples. The amount of phase impurities that can be therein sometimes too small is to be detected by the most widely used methods of characterization (XRD, FTIR, etc.).

Very recently, it has been found that Eu²⁺-doped strontium aluminate $SrAl_4O_7$ samples exhibit bluegreen persistent luminescence at 490 nm while the co-doping with Dy^{3+} shifts the maximum of emission to 475 nm and, consequently, the colour towards the blue spectral range [6]. It is well known that the emission of Eu²⁺ is very strongly dependent on the host lattice and can occur from the ultraviolet to the red region of the electromagnetic spectrum [4, 7, 8]. The purposes of the present work were to

examine the phase purity and to elucidate the possibility and the extent of structural changes in $\mathrm{SrAl_4O_7}$ environment upon doping and co-doping with $\mathrm{Eu^{2+}}$ and $\mathrm{Dy^{3+}}$ ions using solid-state nuclear magnetic resonance (NMR), namely $^{27}\mathrm{Al}$ MAS (magic-angle spinning) NMR spectra. These complementary data can be useful in trying to explain the blue shift observed in the $\mathrm{SrAl_4O_7}$: $\mathrm{Eu^{2+}}$ emission spectra upon co-doping with $\mathrm{Dy^{3+}}$.

2. Experiment

The undoped, 3% Eu-doped and 6% Dy-co-doped $SrAl_4O_7$ samples were prepared by the solid state-reaction method. The phase purity of synthesized specimens was examined by powder X-ray diffraction measurements. The details of chemical preparation, synthesis and characterization are presented in Ref. [6].

The solid-state 27 Al MAS NMR experiments were performed using a 400 MHz *Bruker* AVANCE III HD spectrometer with a 4 mm-wide bore double resonance HX CP-MAS probe. The experiments were performed in 9.4 T magnetic field using an Ascend wide bore superconducting magnet. The Larmor frequency for 27 Al was 104.3 MHz and chemical shifts were referenced to 1 M Al(NO₃)₃. The MAS frequency (v_{MAS}) was 12 kHz. 27 Al MAS spectra were measured applying the single pulseacquire sequence using the short duration (2 μ s) excitation pulse, the number of scans was 512 and the recycle delay was 10 s.

In addition, the ²⁷Al MAS spectra were acquired with the rotor synchronized Hahn-echo sequence $(\pi/6 - \tau - \pi/3 - \tau - \text{acq})$ [9], setting $\tau = 10/v_{\text{MAS}}$ (Fig. 1).

The NMR spectra were processed using the Topspin 3.2 software. Some additional processing was carried out using the Microcal Origin 9 package.

3. Results and discussion

The structure of SrAl₄O₇ has been studied in several works [10-12]. The structure was deduced to be a monoclinic structure with the space group C12/c1 [10, 11]. The $SrAl_4O_7$ lattice consists of corner sharing AlO₄ tetrahedral layers and strontium ions, which are embedded in between the layers [2, 13]. Layers of highly interlinked AlO₄ tetrahedra extend in the *xy* plane and give rise to the enhanced growth speed in the *x* and *y* direction. In between the layers, one finds the Sr ions as well as the AlO₆ octahedra that act as bridges [13]. Each Sr atom in SrAl₄O₇ is surrounded by ten O atoms with interatomic distances from 2.50 to 2.79 Å [2]. This compound also has a high-pressure form β -SrAl₄O₇. The crystal lattice of this form consists of a 3-dimensional network of Al(1)O₆ octahedra, and A1(2)O₄ and A1(3)O₄ tetrahedra. The A1-O bond lengths were found to be 1.795 to 1.968 Å for the octahedra and 1.449 to 1.537 Å for the tetrahedra [6]. These distances in A1O4 units are considerably shorter than those in other aluminates.

In literature there are known several synthesis routes used for the preparation of $\mathrm{SrAl_4O_7}$. They are reviewed in Ref. 6. However, the route of conventional solid-state reactions is seldom reported. Moreover, in some works it was claimed that $\mathrm{SrAl_4O_7}$ cannot be synthesized by the solid-state reaction. In our earlier paper [6], a successful synthesis of $\mathrm{SrAl_4O_7}$ samples doped with $\mathrm{Eu^{2+}}$ and co-doped

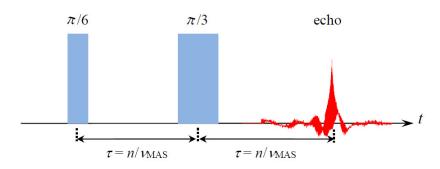


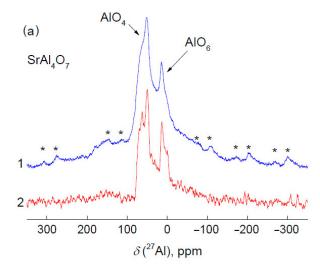
Fig. 1. The rotor synchronized Hahn-echo pulse sequence used in 27 Al MAS NMR experiments. The value n=10 was chosen for all experiments in the present work.

with Dy³⁺ by the conventional solid-state reaction method was carried out.

A broad variety of solid-state NMR techniques, such as ²⁷Al MAS, multiple quantum magic-angle spinning (MQ-MAS) and heteronuclear multiple quantum coherence (HMQC) spectroscopy, have been effectively applied elucidating very fine structural details in series of ions-doped aluminates [1, 14–19]. Therefore, in the present work the ²⁷Al MAS NMR was applied in order to complementarily characterize the synthesized SrAl₄O₇ using the conventional solid-state reaction method as well as 3% Eu-doped and 6% Dy-co-doped SrAl₄O₇.

As the major factor that controls the ²⁷Al chemical shifts $\delta(^{27}\text{Al})$ is the atomic structural environments of aluminum atoms, i.e. the nearest-neighbour coordination geometry, the ²⁷Al MAS NMR spectra provide useful, sometimes even unique information on Al sites and coordination numbers. However, not always all of the ²⁷Al resonances corresponding to different Al sites are well resolved in the spectra. Going through literature data [4, 18, 19 and Refs. cited therein], it can be stated that in most Al-O environments, the ²⁷Al NMR peaks of 6-coordinated Al (AlO6) appear in the range covering -20 to +15 ppm and are well separated from the signals from the 4-coordinated Al (AlO₄) that appear at 50 to 145 ppm. It looks that the interchange of ions $Sm \leftrightarrow Cs \leftrightarrow Sr$ in various aluminates does not have too much influence on this sequence and the range of variation of ²⁷Al chemical shifts.

The ²⁷Al MAS NMR spectra of pristine as well as 3% Eu-doped and 6% Dy-co-doped SrAl₂O₂ samples studied in the present work are shown in Fig. 2. The Hahn-echo pulse sequence is typically used for solid-state NMR measurements when the single-pulse spectrum experiences a strong background signal due to, e.g. long probe dead time or ringing effects. By changing echo time it could also be used as a relaxation filter, in other words, fast relaxing spectral components could be filtered out if two or more coexisting species, which relax at different rates, are present in the sample. Note that a higher resolution and reduction of spinning sideband intensities were achieved applying the Hahn-echo pulse sequence in the present case (Fig. 2(a)). The peaks of 4- and 6-coordinated Al are well resolved and easily recognized in the spectra. No significant changes in the ²⁷Al chemical shifts upon Eu²⁺ and Dy³⁺ dop-



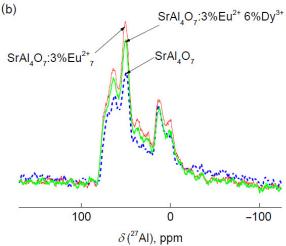


Fig. 2. ²⁷Al MAS NMR spectra of pristine $SrAl_4O_7$ ((a): 1 for single-pulse, 2 for Hahn-echo experiment, respectively), and 3% Eu-doped and 6% Dy-co-doped $SrAl_4O_7$ ((b): all Hahn-echo). Asterisks denote spinning sidebands.

ing are seen. Only the relative intensities can be noted, and they are more pronounced for 4-coordinated Al (Fig. 2(b)). A very asymmetric shape of AlO₄ signals with the tails to lower values of ²⁷Al chemical shifts point towards the presence of a local disorder and thus a broad distribution of quadrupolar coupling constants and chemical shifts.

It is known that the 5-coordinated Al (AlO₅) yields $\delta(^{27}\text{Al})$ in the range +15 to +35 ppm. This was observed in the ^{27}Al NMR spectrum of glassy YAG-4Si, however not seen in the pure crystalline YAG and in the spectra of annealed YAG-4Si [18]. It means that this spectral feature can be used evaluating the structural disorder and phase purity in the sample. However, note that

a certain polemic around the existence and detection of the $^{27}\mathrm{Al}$ NMR signals from AlO_5 sites run in the literature [16, 17, 19]. The AlO_5 signal, if it is at all present in the aluminate under investigation, is often obscured by the overlap with strong AlO_6 signals. The most powerful technique to reveal the presence of AlO_5 is MQ-MAS. The study applying this technique on the synthesized materials is in progress.

4. Concluding remarks

- 1. The 27 Al MAS NMR experiments have shown that the Eu²⁺ with Dy³⁺ ion doping carried out using the solid state-reaction method did not affect the bulk structure and the local Al environment in SrAl₄O₇.
- 2. The spectral features observed in the range between 4- and 6-coordinated Al peaks (viz. 20–40 ppm) demand one to recognize that certain phase imperfections are present, most probably the amorphous/glassy domains appear. Also note that the chosen solid state-reaction regime does not alter it, i.e. the spectral features that supposedly may reflect the phase purity were the same in all studied samples.
- 3. The ²⁷Al MAS NMR technique appeared to be a very effective tool for the analysis of phase perfectness. Such amount of phase impurities was too small to be detected by widely used XRD or FTIR methods [6].
- 4. Large shifts of the emission maximum (from 490 to 475 nm) by co-doping of SrAl₄O₇:Eu²⁺ with Dy³⁺ cannot be caused by the local environment changes upon the doping. Since the main factor governing the emission of Eu²⁺ is the crystal field surrounding the ion, one possible explanation might be that the charge disbalance induced by 3⁺ ions influences the strength of the crystal field around Eu²⁺ (without noticeably affecting the structure).

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Eu²⁺ LEGIRUOTŲ IR Dy³⁺ KOLEGIRUOTŲ SrAl₄O₇ TYRIMAI ²⁷Al MAS BMR SPEKTROMETRIJOS METODU

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Santrauka

Eu²+legiruoti stroncio aliuminatai SrAl₄O₂ pasižymi mėlynai žalia išliekančiąja liuminescencija ties 490 nm, o juos kolegiravus Dy³+ jonais emisijos maksimumas pasislenka link 475 nm. Kietafazių reakcijų metodu buvo paruošti nelegiruoti, 3 % Eu legiruoti ir 6 % Dy kolegiruoti SrAl₄O₂ mėginiai, kurie ištirti taikant kietojo kūno ²¬Al MAS ("magiško kampo sukimo") BMR spektrometrijos metodą. Naudotos pavienio impulso ir Hahno sukinių aido impulsų sekos. Nustatyta, kad SrAl₄O₂ legiravimas Eu²+ ir Dy³+ jonais neturi įtakos kristalinei struktūrai visame bandinio tūryje bei lokaliai Al atomų aplinkai. Tai reiškia, kad toks pastebimas emisijos maksimumo poslinkis SrAl₄O₂;Eu²+ kolegiruo-

jant Dy³+ nėra nulemtas lokalios kristalinės struktūros pokyčių. Kita vertus, BMR spektro dalis, stebima tarp keturių ir šešių koordinuotų Al signalų (20–40 m.d.), parodo, kad šiuose bandiniuose yra kitų fazių priemaišų, ir, labiausiai tikėtina, amorfinių / stikliškų domenų. Pažymėtina, kad tokie maži fazinių priemaišų kiekiai nebuvo aptikti įprastais ir bene dažniausiai šiam tikslui pasiekti taikomais rentgeno difrakcijos (XRD) bei Furjė vaizdavimo infraraudonosios spektrometrijos (FTIR) metodais. Tai rodo, kad ²¬Al MAS BMR metodas gali būti labai efektyvus įrankis kontroliuojant stroncio aliuminatų serijos junginių fazinį grynumą.